# Ion-Paired Amphiphiles (Final Report)

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### Statement of The Problem

Amphiphilic ion pairs, derived from single-chain cations and single-chain anions (i.e., ion-paired amphiphiles, IPA's) were previously introduced as novel vesicle-forming materials. The primary objective of this program has been to define the scope of such IPA's, via a careful examination of molecular structure--supramolecular structure/property relationships. Related studies were aimed at exploring the feasibility of creating polymerized vesicles from IPA's. A secondary objective of this program is to examine the feasibility of using IPA-based vesicles as sensors for ionic species present in soluton, and as actuators for triggering the disruption of conventional double-chain surfactant vesicles. The ultimate goal of this program was to expand IPA chemistry for both theoretical investigations and for applied research.

# Summary of The Most Important Results

Work that was carried out under this program demonstrated that a wide variety of amphiphilic ion pairs can function as novel membrane-forming materials. In a broad sense, this work has significantly expanded the scope of synthetic surfactants for use in the membrane area. The demonstration that ionically-paired single chain surfactants can produce bilayer structures is particularly significant because it bridges the gap between single- and double-chain amphiphiles in terms of molecular structure-aggregation activity relationships. In addition, the finding that counterions can play a major role in defining the permeability of surfactant bilayers highlights the fact that there is considerably more room for fine-tuning the barrier properties of resulting membranes than had previously been realized.

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#### **Publications**

All of the work that has been carried out during the course of this project have been published:

- 1) Y. Chung, S. L. Regen, Macromolecules, 24, 5738 (1991)
- 2) K. Hirano, H. Fukuda, S. L. Regen, Langmuir, 7, 1045 (1991).
- 3) Y. Chung, H. Fukuda, K. Hirano, S. L. Regen, Langmuir, 8, 2842 (1992).
- 4) Y. Chung, S. L. Regen, Langmuir, 9, 1937 (1993).
- 5) S. Watanabe, S. L. Regen, J. Am. Chem. Soc., 116, 5762 (1994).
- 6) S. Watanabe, S. L. Regen, J. Am. Chem. Soc., 116, 8855 (1994).

## Participating Scientific Personnel

- 1) S. L. Regen (principal investigator)
- 2) Dr. Yong Chung (postdoctoral fellow)
- 3) Dr. Shinji Watanabe (postdoctoral fellow)

#### Inventions

none

# REPORT DOCUMENTATION PAGE

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#### 13. ABSTRACT (Maximum 200 words)

Work that was carried out under this program demonstrated that a wide variety of amphiphilic ion pairs (i.e., ion-paired amphiphiles, IPA's) can function as novel membrane-forming materials. In a broad sense, this work has significantly expanded the scope of synthetic surfactants for use in the membrane area. The demonstration that ionically-paired single chain surfactants can produce bilayer structures is particularly significant because it bridges the gap between single- and double-chain amphiphiles in terms of molecular structure-aggregation activity relationships. In addition, the finding that counterions can play a major role in defining the permeability of surfactant bilayers highlights the fact that there is considerably more room for fine-tuning the barrier properties of resulting membranes than had previously been realized.

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